|    | solution of dichromic acid is mixed with a solution of sodium carbonate |
|----|---|
|    |   |
|    | is what the gas does to a glowing splint.                               |
| S  | olid manganese (VII) chlorate decomposes when it is heated.             |
|    | is what the gas does to a glowing splint.                               |
| A  | mmonia gas and carbon dioxide gas are bubbled through water.            |
|    | is the number of spectator ions.  |
| G  | aseous ethyne burned in limited oxygen.                                 |
|    | is the organic compound.  |
| A  | mmonium gas is bubbled through a suspension of aluminum oxalate.        |
|    | is the approximate pH of the resulting solution.                        |
| C  | alcium turnings are added to water.                                     |
|    | is what the gas does to a glowing splint.                               |
| A  | crystal of sodium acetate is dropped into water.                        |
|    | is what this salt does to water.  |
| So | olid copper (III) oxide is added to water.                              |
|    | in the man and describe   |
|    | is the gas produced.  |

| 9)  | Carbon dioxide gas is bubbled through a solution of sodium phosphate.   |
|-----|---|
| 10) | is the name of the acid produced.  Gaseous hydrogen sulfide is passed through a solution of potassium fluoride.         |
| 11) | is a spectator ion.  Carbon dioxide gas is passed over solid barium oxide.  |
| 12) | is what happens to the oxidation number of the barium.  A piece of magnesium ribbon is dropped into dilute nitric acid. |
| 13) | is the oxidation number of N in the nitric acid.  A piece of magnesium ribbon is dropped into concentrated nitric acid. |
| 14) | is the color of the evolved gas.  A piece of magnesium ribbon is dropped into dilute sulfuric acid.                     |
| 15) | is the color of the evolved gas.  A piece of magnesium ribbon is dropped into concentrated sulfuric acid.               |
| 16) | is the oxidation number of S in sulfuric acid.  Fluorine gas is bubbled through a solution of lithium iodide.           |
|     | is the species that is the reducing agent.  |

| 17) | Solid lead (II) hydroxide is added to a solution of ammonium chromate. |
|-----|--|
|     | is the gas produced.   |
| 18) | Solid calcium bicarbonate is added to a solution of potassium sulfate. |
|     | is a spectator ion.  |
| 19) | Solutions of aluminum acetate and sodium chlorite are mixed.           |
|     | is a spectator ion.  |
| 20) | Oxygen gas is passed over solid aluminum chloride.                     |
|     | is the oxidation number of the O in the oxygen gas.                    |